

#### Review article

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Present status (PS) of integrated bio refineries (IBRs) and sustainable development engineering (SDE) together with editorial of future research (FR) for multi-disciplinary (MD), multi-scales (MSs), multi-plat-forms (MPFs), multi renewable raw materials (MRRMs) feed-stocks (FSs) IBR for SDE

#### Abstract

SDE is more advanced than Environmental Engineering (EE), the best way to analyze their relations is by using the System Theory (ST) especially the Integrated System Approach (ISA) which from a certain point of view shows EE as a subsystem of SDE with the other most important sub-system being RRMs.

From another point of view SDE can be considered a subsystem of Sustainable Development (SD), with the other sub-systems being the other sustainability sub-systems, e.g.: SD-economics; SD-politics; SD-sociology; SD-production; SD-consumption; SD-ethics; etc.

IBRs are the most important sub-systems of SDE especially with regards to production from MRRMs. This Editorial Paper (EP) introduces PS IBRs and the future ones with large number of MRRM FSs and larger number MPFs.

**Keywords:** renewable raw materials, multiple RRMS, integrated bio refineries, feed stocks, bio fuels, bio products, bio energies, global warming, multi-dimensional, multi cross disciplinary, carbon nano tubes, lignocellulose, lignin, lipid, microalgae, carbohydrates, biomass conversion, environmental engineering, sustainable development, SD engineering, CO<sub>2</sub> emission, CO<sub>2</sub> sequestration, trans-esterification, photo bio reactors, biotechnology, nanotechnology, bioethanol, biodiesel, fermentation, membrane fermenters, valorization, de-valorization

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**Abbreviations:** RRMs, renewable raw materials; MRRMs, multiple RRMS; IBRs, integrated bio refineries; FSs, feed stocks; BFs, bio fuels; BPs, bio products; BEs, bio energies; GW, global warming; MD, multi-dimensional; MCD, multi cross disciplinary; CNTs, carbon nano tubes; BC, biomass conversion; EE, environmental engineering; SD, sustainable development; SDE, SD engineering; MFs, membrane fermenters

# Introduction

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This EP addresses in an integrated manner, the development of IBRs, SD and their relation to each other. With special emphasis on the use of RRMs to achieve SD and not only a clean environment. The article addresses all distinction between Environmental Engineering (EE) and Sustainable Development Engineering (SDE).

Many parts address the MRRMs and their choice as well as the suitable enzymes and microorganisms to be used.

The heart of the paper and most of its parts will address the challenges of moving from the present two platforms and one RRM to the complex novel ones with MPFs and MRRMs FSs. It presents the advancement towards MD IBRs with a variety of products, e.g.: BFs, BPs, Bio-polymers and Bio-medicine from a wide range of MRRMs. It addresses EE and SDE using simple IBRs with two PFs as well as complex ones with MRRMs FSs and large number of PFs producing a wide range of BFs and BPs including Carbon Nano Tubes (CNTs)

from lignin extracted from lignocellulose waste materials, as well as Bio-Energies (BEs). The treatment of wide range of MRRMs to produce a wide range of BFs; BPs and BEs will fight pollution, Global Warming (GW) and  $CO_2$  emission, etc.

SD is formed of a large number of subsystems such as the engineering one (SDE), economical ones, etc. Therefore, the subject is MD by its very nature. This EP concentrates on SDE which, from a ST point of view, can be formed of the two subsystems: EE and MRRMs.<sup>1-5</sup> From a Non-linear Dynamics (NLDs) point of view, especially stability theorem, it can be divided into necessary but not sufficient condition which, in the process industry, is namely: Maximum Production Minimum Pollution (MPMP), what makes this necessary condition, sufficient is the introduction of RRM for simple IBRs and MRRMs for the complex MD MPFs IBRs.<sup>6</sup> A relatively simpler analysis of these complex IBRs has been presented earlier.<sup>7.8</sup>

IBRs are the most important integrated tools to achieve SD with regard to BFs and BPs, particularly when using MRRMs. The PS IBRs process biomass fed through two platforms, one biochemical and the other thermal; the single platform is not an IBR but an Elementary Bio-Refinery (EBR).<sup>9</sup> However, IBRs still require extensive development with regards to its configuration, the optimization of each unit, and the overall efficiency of the processes. What's more, two platforms IBRs are not the end of the story; more platforms using different kinds of biomass waste, energy crops, or microalgae (natural and synthetically

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produced from photo-bioreactors using CO<sub>2</sub> as feedstock) are possible. These multi-platforms IBRs have the potential to drastically increase the CO<sub>2</sub> consumption of IBRs by utilizing natural and industrially generated CO<sub>2</sub>. They also expand the range of raw materials that can be employed, for example agricultural wastes such as rice, wheat straw, cotton stacks, and corn Stover (which are mainly lingo-cellulose) also energy crops such as switch grass (which is mainly lingo-cellulose) and Jatropha (which is a mixture of lingo-cellulose and lipid) can be used. In addition to algae (natural and produced from CO<sub>2</sub> in photo bioreactors) that can also be used.

IBRs and this suggested MD IBRs are becoming most important after nature did ring the bell of anger during this destructive wave of Global Warming (GM) which is a negative subsystem of the neglect of SD.

### **Goals and discussion**

This editorial paper addresses IBRs both in the PS and future novel developments. A specific focus is on the application of MRRMs in IBRs of different degrees of complexity and sophistication with regard to the configuration, ensuring sustainability by replacing non-RRMs by RRMs. When the raw material is changed, it is typically accompanied by a change in technology, and this opens the door for advanced research using Mathematical Modeling (MM), Computer Simulation (CS) and Experimental Verification (EV) to integrate chemical processes with biochemical techniques to achieve the optimal synergy of the two or more platforms IBRs, this will also involve the use of novel technologies such as membrane reactors.<sup>6-9</sup> The overall aim is to be able to produce a wide range of BFs and BPs from a wide range of MRRMs in MPFs IBR using novel technologies. The analysis of an IBR includes that of its optimal configuration, simple process optimization using Mass and Heat Balances (MHBs), and more advanced optimization using MM, CS, EV and Computational Fluid Dynamics (CFD) techniques in a unified manner (e.g.: physico-chemical or empirical models are created and then their accuracy is improved in a feedback loop with experimental data). Such optimization for each RRM is essential to achieve maximum efficiency of the particular IBR, while at the same time providing flexibility to convert any feedstock into power, heat and value-added products (BFs, BPs) in a sustainable manner. In addition, each subsystem of each platform is investigated and optimized with

Table I Numbers of PFs for different types of MD, MSs, MRRMs FSs for SDE

regard to its configuration, optimal design and optimal mode of operation taking into consideration modern findings in both sides, e.g.: membrane reactors and bio-reactors, integrated auto-thermic reactors, exploration of the implications of bifurcation and chaos in autonomous and non-autonomous systems, etc.<sup>10,11</sup>

Many articles and review ones discuss the different approaches for dealing with IBRs, especially with regard to the use of novel technologies, e.g.: membrane catalytic and bio-catalytic reactors, in addition to other challenges associated with BFs, BPs and BEs from RRMs as well as other sides of IBRs.<sup>12-20</sup>

In addition to the above, the number of platforms can be increased with increasing the number of BPs using the same nonedible MRRMS. For example, additional platforms can be added by using the same MRRMs biomass to produce Bio-polymers and Biomedicines<sup>21,22</sup> from the same MRRMs, we will start in this article by 3 Bio-polymers and 3 Biomedicines. Therefore additional 6 platforms are added per one RRM. In addition, the main biochemical platform used to produce BFs/BPs from the lingo-cellulosic non-edible RRM biomass, the first step is the delignification of lingo-cellulose to separate lignin in one side from cellulose-hemicellulose in the other side. The present state is that lignin is completely used to produce energy and electricity for the process and they are more than what is needed by the process and extra electricity is sold. What is suggested in this EP is to create an additional platform to process the extra lignin to produce Carbon Nanto Tubes (CNTs) in a special Fluidized bed with Chemical Vapor Deposition (CVD) catalyst.<sup>23-27</sup> The thermal platform can be made into 2 one is the usual gasification to syngas after conditioning is reacted in Fischer Tropsch reactors to produce BFs and BPs. The second is pyrolysis to produce bio-oil which is cracked in a Fluid Catalytic Cracking (FCC) unit to produce a range of BFs. Therefore, according to the above the total number of platforms per RRM of the first 4 RRMs is 10 PFs. For the other 3 RRMs (i.e.: Jatropha and Algae) PFs=11 per RRM taking into consideration a PF for separation of lipid and transforming it to biodiesel. Therefore for this MD, MSs, MPFs, MRRMs FSs for SDE with 4 standard MRRMs Lignocellulose biomass + one Jatropha and 2 algae the total number of PFs is =4X10+3X11= 73 PFs for this MD IBR. For PS IBR with one RRM FS the total number of PFs is 2; the IBRs in between are given in Table 1 below.

Type of IBR	Total no. of MRRMs, NT=NLGC+ NJA	No. of Lignocellulose, (LGC) MRRMs , NLGC	No. of Jatropha and Algae MRRMs, NJA	No. of Platforms: NPFs=NLGC*10 +NJA*11 Except for the PS simple IBRs with NT=1, NPFs=2		
1	1		0	2		
7	7	4	3	73		
2	2	2	0	20		
3	3	3	0	30		
4	4	4	0	40		
5	5	4	I	51		
6	6	4	2	62		
8 Additional example 1 (e.g.1)	8	5	3	83		
e.g.2	9	5	4	94		
e.g. 3	11	6	5	115		

Applying material balances in a simplified systematic approach<sup>28</sup> is shown below in sequential description then summarized in Table 2, The lignin  $L_i$  can be divided into two parts for each i = 1-7 as shown

below. The fraction of  $L_i$  to be transformed to Benzene is  $B_i$  and the rest is transformed to CNTs. The choice of the  $B_i$  values depends upon economics and market conditions.

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- I. Biomasses, i=1-4, the following sequence of relations valid for all 4 first RRM, not for Jatropha and the 2 types of algae:
  - 1) Lignocellulose in any  $RRM_i \equiv LC_i$ , i=1-4
  - 2) Lignin in all RRMs except  $\equiv L_i$ , i=1-4
  - 3)  $\mathbf{x}_i = \mathbf{L}_i / \mathbf{LC}_i$  (LC<sub>i</sub> is RRM<sub>i</sub>) $\rightarrow$ L<sub>i</sub> =  $\mathbf{x}_i \cdot \mathbf{LC}_i$ , i=1-4
  - 4) Carbon Nano Tubes from different  $L_i \equiv CNTs_i$  i=1-4
  - 5)  $a_i = CNTs_i / L_i \rightarrow CNTs_i = a_i \cdot L_i = a_i \cdot X_i \cdot LC_i (LC_i \text{ is } RRM_i), i=1-4$
  - 6)  $B_i \equiv$  fraction of  $L_i$  used to produce Benzene ( $C_6 H_6$ )

#### II. Jatropha i=5:

- 1) (Ligno-cellulose, LC<sub>5</sub>)/ (total Jatropha, LC<sub>T5</sub> (which is RRM<sub>5</sub>)) = $y_5 \rightarrow LC_5 = y_5$ , LC<sub>T5</sub> =  $y_5$ , RRM<sub>5</sub>
- 2) Lignocellulose in any Jatropha  $\equiv$  LC<sub>5</sub>
- 3) Lignin in lignocellulose of Jatropha  $\equiv$  L<sub>5</sub>
- 4)  $x_5 = L_5 / LC_5 \rightarrow L_5 = x_5. LC_5$
- 5) Carbon Nano Tubes from  $L_5 \equiv CNTS_5$
- 6)  $a_5 = CNTs_5 / L_5 \rightarrow CNTs_5 = a_5 \cdot L_5 = a_5 \cdot x_5 \cdot LC_5$
- 7)  $B_5 \equiv$  fraction of  $L_i$  used to produce Benzene ( $C_6 H_6$ )

#### III.Algae (natural):

- 1) Lignocellulose in natural algae  $\equiv LC_6$
- 2) (Lignocellulose in natural algae,  $LC_6$ )/ (total natural algae,  $LC_{T6}$ (which is RRM<sub>6</sub>)) =  $y_6 \rightarrow LC_6 = y_6$ .  $LCT_6 = y_6$ . RRM<sub>6</sub>
- 3) Lignin in lignocellulose of natural algae  $\equiv L_6$
- 4)  $x_6 = L_6 / LC_6 \rightarrow L_6 = x_6 \cdot LC_6$
- 5) Carbon Nano Tubes from  $L_6 \equiv CNTS_6$
- 6)  $a_6 = CNTs_6 / L_6 \rightarrow CNTs_6 = a_6 \cdot L_6 = a_6 \cdot x_6 \cdot LC_6$
- 7)  $B_6 \equiv$  fraction of Li used to produce Benzene (C<sub>6</sub> H<sub>6</sub>)

Table 2 Of the state variables, physical, design and optimization parameters

8) Approximate estimate of Lipid in Algae,  $LP_6 = (1-y_6)$ . LCT<sub>6</sub> =  $LP_6 = (1-y_6)$ . RRM<sub>6</sub>

#### IV. Algae, synthetic, e.g., in a photo-bioreactor with CO, feed:

- 1) Lignocellulose in synthetic algae, e.g., in a photo-bioreactor with CO<sub>2</sub> feed  $\equiv$  LC<sub>2</sub>
- Lignocellulose in synthetic algae, LC<sub>7</sub>)/ (total synthetic algae, LCT<sub>7</sub> (which is RRM<sub>7</sub>)) =y<sub>7</sub> →LC<sub>7</sub>= y<sub>7</sub>. LCT<sub>7</sub> = y<sub>7</sub>. RRM<sub>7</sub>
- 3) Lignin in lignocellulose of natural algae  $\equiv L_{\gamma}$
- 4)  $\mathbf{x}_{\gamma} = \mathbf{L}_{\gamma} / \mathbf{L}\mathbf{C}_{\gamma} \rightarrow \mathbf{L}_{\gamma} = \mathbf{x}_{\gamma} \cdot \mathbf{L}\mathbf{C}_{\gamma}$
- 5) Carbon Nano Tubes from  $L_7 \equiv CNTs_7$
- 6)  $a_7 = CNTs_7 / L_7 \rightarrow CNTs_7 = a_7 \cdot L_7 = a_7 \cdot x_7 \cdot LC7$
- 7)  $B_{\gamma} \equiv$  Benzene (C<sub>6</sub> H<sub>6</sub>) from  $L_{\gamma} = L_{\gamma} (a_{\gamma}, x_{\gamma}, LC_{\gamma}) = (x_{\gamma}, LC_{\gamma}) (a_{\gamma}, x_{\gamma}, LC_{\gamma}) = (1 a_{\gamma}). (x_{\gamma}, LC_{\gamma})$
- 8) Approximate estimate of Lipid in Algae,  $LP_{\gamma=}(1-y_{\gamma})$ .  $LCT_{\gamma} = (1-y_{\gamma})$ .  $RRM_{\gamma}$
- 9) Approximate estimate of  $CO_2$  consumed per ton Algae produced  $\equiv Y_{CDOIDA}$
- 10)  $CO_2$  consumption =  $Y_{CDOtoA}$ .  $LCT_7 = Y_{CDOtoA}$ . RRM<sub>7</sub>

This benzene above is reacted with a part z of the bioethanol resulting from the cellulose -hemicellulose in the above 7 steams =

$$B = (B_1 + B_2 + B_3 + B_4 + B_5 + B_6 + B_7)$$
.  $z = z$ .  $\sum_{i=1}^{n} B_i$ , This Benzene if produced

from certain fractions of  $L_i$  will be reacted with Bioethanol produced from the biochemical platform to produce Ethyl Benzene (EB) which is then dehydrogenated to Styrene and then polymerized to Polystyrene used in the foam industry. Most efficient modern efficient processes should be used, e.g.: relatively advanced process for dehydrogenation of EB to styrene is the catalytic process with hydrogen selective membranes to beak the thermodynamic equilibrium barrier of this reversible reaction and also producing pure hydrogen as a secondary product (Table 2).<sup>29</sup>

	RRM <sub>:</sub> Renewable raw material. tons/day	LC <sub>.</sub> , Ligno- cellulose	LP,, Lipid	LC <sub>i</sub> +LP <sub>i</sub>	x <sub>i</sub> =L/ LC <sub>i</sub> =fraction Lignin in Ligno- cellulose	L <sub>i</sub> , Lignin= x <sub>i</sub> . LC <sub>i</sub>	Cellulose- Hemicellulose CHC <sub>i</sub> = LC <sub>i</sub> - L <sub>i</sub>	Carbon Nano Tubes CNTs <sub>i</sub> , tons/ day	a,= CNTs,/ L,, fraction of CNTs to Lignin production ratio	y;= fraction LC <sub>i</sub> / RRM <sub>i</sub>	B <sub>i</sub> , fraction of L <sub>i</sub> made into C <sub>6</sub> H <sub>6</sub>
Biomass I	RRM	LC <sub>1</sub> = RRM <sub>1</sub>	LP <sub>1</sub> =0	LC <sub>1</sub> =RRM <sub>1</sub>	L <sub>I</sub> /LC <sub>1</sub>	L <sub>i</sub> =x <sub>i</sub> .LC <sub>i</sub>	CHC <sub>I</sub> = LC <sub>I</sub> - L <sub>I</sub>	CNTs <sub>1</sub>	CNTs <sub>1</sub> /L <sub>1</sub>	I	B
Biomass 2	RRM <sub>2</sub>	$LC_2 = RRM_2$	LP <sub>2</sub> =0	$LC_2 = RRM_2$	L <sub>2</sub> /LC <sub>2</sub>	$L_2 = x_2 LC_2$	CHC <sub>2</sub> = LC <sub>2</sub> - L <sub>2</sub>	CNTs <sub>2</sub>	CNTs <sub>2</sub> /L <sub>2</sub>	I	B <sub>2</sub>
Biomass 3	RRM <sub>3</sub>	$LC_3 = RRM_3$	LP <sub>3</sub> =0	LC3=RRM3	L <sub>3</sub> /LC <sub>3</sub>	L <sub>3</sub> = x <sub>3</sub> .LC <sub>3</sub>	CHC3= TC3- T3	CNTs <sub>3</sub>	CNTs <sub>3</sub> /L <sub>3</sub>	I	B <sub>3</sub>
Biomass 4	RRM <sub>4</sub>	$LC_4 = RRM_4$	LP <sub>4</sub> =0	$LC_4 = RRM_4$	L <sub>4</sub> /LC <sub>4</sub>	$L_4 = x_4.LC_4$	$CHC_4 = LC_4 - L_4$	CNTs <sub>4</sub>	CNTs <sub>4</sub> /L <sub>4</sub>	I	$B_4$
Jatropha	RRM <sub>s</sub>	LC <sub>s</sub> = y <sub>s</sub> . RRM <sub>s</sub>	LP <sub>s</sub> = (I- y <sub>s</sub> ). RRM <sub>s</sub>	LC <sub>s</sub> +LP <sub>s</sub>	L <sub>s</sub> /LC <sub>s</sub>	L <sub>s</sub> = x <sub>s</sub> .LC <sub>s</sub>	$CHC_s = LC_s - L_s$	CNTs <sub>5</sub>	CNTs <sub>5</sub> /L <sub>5</sub>	LC <sub>s</sub> / RRM <sub>s</sub>	B <sub>s</sub>
Natural Algae	RRM <sub>6</sub>	LC <sub>6</sub> = y <sub>6</sub> . RRM <sub>6</sub>	LP <sub>6</sub> = (I- y <sub>6</sub> ). RRM <sub>6</sub>	LC <sub>6</sub> +LP <sub>6</sub>	L <sub>6</sub> /LC <sub>6</sub>	L <sub>6</sub> = x <sub>6</sub> . LC <sub>6</sub>	CHC <sub>6</sub> = LC <sub>6</sub> - L <sub>6</sub>	CNTs <sub>6</sub>	CNTs <sub>6</sub> /L <sub>6</sub>	LC <sub>6</sub> / RRM <sub>6</sub>	B <sub>6</sub>
Synthetic Algae	RRM <sub>7</sub>	LC <sub>7</sub> = y <sub>7</sub> . RRM <sub>7</sub>	LP <sub>7</sub> = (I- y <sub>7</sub> ). RRM <sub>7</sub>	LC <sub>7</sub> +LP <sub>7</sub>	L <sub>7</sub> /LC <sub>7</sub>	L <sub>7</sub> = x <sub>6</sub> . LC <sub>6</sub>	CHC <sub>7</sub> = LC <sub>7</sub> - L <sub>7</sub>	CNTs <sub>7</sub>	CNTs <sub>7</sub> /L <sub>7</sub>	LC <sub>7</sub> / RRM <sub>7</sub>	B <sub>7</sub>

**CNTs:** Molecular Formula of Lignin:  $C_{18}H_{13}N_3Na_2O_8S_2$ , i.e.: 1 mole of  $L_i$  gives 18 moles of C. Molecular weight of Lignin= 509.4, i.e.: 509.4 gm lignin will give 18 X12= 216 gm Carbon, 1 gm Lignin = 216/509 = 0.424 gm Carbon. Thus, the values of  $a_i$  based on 100%

conversion ( $X_{conv} = 1.0$ ) =0.35-0.45 depending on the type of  $L_i$  and type of CNTs<sub>i</sub>. The actual conversion  $X_{act}$  will depend on the State of the Art (SA) of the design and optimization of Fluidized Bed Reactor with Chemical Vapor Deposition (CVD) Catalyst for converting  $L_i$ 

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to  $\text{CNTs}_{i}$ . The SA may allow  $X_{\text{act}} = 0.4-0.5$ . Further research using kinetics model as well as reactors modeling, experimental verification and will increase  $X_{\text{act}}$  considerably.<sup>30</sup>

**B**<sub>i</sub>: These are the values of the fraction of fraction of  $L_i$  to be used to produce  $C_6H_6$  and are arbitrary depending upon economics and market conditions.

 $CO_2$  consumption for synthetic algae: Consumption of  $CO_2$  emitted which contributes positively against

 $GW = Y_{CDOtoA}$ .  $LCT_7 = Y_{CDOtoA}$ .  $RRM_7$ 

**x<sub>i</sub> and y<sub>i</sub>:** The x<sub>i</sub>, which is the fraction of lignin in the lignocellulose, vary depending on the type of biomass, its strain and whether the lignocellulose is almost the whole biomass i=1-4, or it is part of Jatropha or algae, i=5-6. It varies in the range ~ 0.15- 0.30 for i=1-4,<sup>31</sup> for i=5-7 it varies in the range ~ 0.15-0.20.<sup>32</sup> For i=1-4, which are free of the lipid y<sub>i</sub> which is the fraction of lignocellulose =1.0 and for i= 5-7 it varies widely in the range ~ 0.25-0.55 depending on the strain of Jatropha and the natural or synthetic (e.g.: in photo bioreactors) algae.<sup>33-36</sup>

This EP is very useful in introducing the audience to the present 2 platforms and one biomass RRM FS and it also introduces the audience to the novel complex MFSs with MPFs IBRs with multi platforms and multiple RRMs feedstock in order to achieve SD, reduction of  $CO_2$  emission and other environmental and economic benefits.

Most papers published are dealing with 2 platforms and one RRM feedstock. This EP will deal with large number of platforms with a large number of MRRMs FSs.

This EP is a clear illustration of the preliminary steps towards the development of Novel(N) Complex(C) IBRs for SD (NCIBRS for SD), to be followed by other steps. It does not only illustrate the use of a wide range of biomasses wastes and algae to produce a wide range of BFs, BPs and BEs that is very useful but it also achieves controllability of CO<sub>2</sub> emission and GW that is beneficial for the earth and its inhabitants. Both aspects are very useful for SD. This stage involves MBs followed by HBs both described by Algebraic Equations (AEs). Followed by the Design Stage (DS) characterized by equations of the types: AEs and Ordinary Differential Equations (ODEs) and Partial Differential Equations (PDEs) for Mass, Heat and Momentum. This stage involves 3 experimental substages:<sup>37–39</sup>

- 1. Gathering rate equations and their parameters for rate processes as well as thermodynamic equilibrium. These can be obtained from the literature, when available, or from own experiments.
- Parameter estimation for verification of all above balances and designs equations against laboratory, pilot plant and commercial (if available) results to ensure the validity of equations for balances and designs calculations.
- 3. Use the above verified equations for sensitivity analysis and optimization for each unit and the entire NCIBR for SD.

Future advanced work will involve the above for dynamic investigation, Direct Digital Control (DDC) and the use of Artificial Intelligence (AI).

As a final comment it should be noticed that many equipment of this NCIBR for SD are affected positively/negatively by static and dynamic bifurcation as well as chaotic behavior. These phenomena should be exploited when their effects are positive and controlled when their effects are negative.<sup>40</sup>

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# **Conflicts of interest**

The authors declare that there are no conflicts of interest.

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