

Comparative phytochemical screening of kenaf and jute leaves

Abstract

This study was conducted to evaluate the comparative phytochemical composition of kenaf and jute leaves. The leaves were harvested at maturity, washed, dried, ground to powder and subsequently subjected to quantitative and qualitative phytochemical screening according to the standard protocols. The results showed that jute leaf contained significantly higher saponin, tannin and glycoside than kenaf leaf. Similarly, jute leaf recorded higher level (25.00mg/100g) of flavonoid than kenaf leaf (20.00mg/100g). In contrast, kenaf leaf recorded slightly higher level of steroid (0.002%), and alkaloid (0.28%) when compared with jute leaf with 0.001% for steroid and 0.25% for alkaloid. Kenaf leaf contained 569.55mg/100g carotenoid while jute leaf had 546.70mg/100g carotenoid. These leaves appeared to be rich in phytochemicals and antioxidants which are good either for human nutrition or medicinal purpose. Screening for the best extraction-solvent showed that water extracted most phytochemicals in both kenaf and jute leaves. Ideally, lipophilic phytochemicals are extracted with protic solvents while hydrophilic phytochemicals efficiently extracted with distilled water in order to retain their physico-chemical properties.

Keywords: Kenaf, phytochemical, jute, solvent

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Introduction

Many leafy vegetables are mainly consumed for their nutritional values without much consideration for their medicinal importance. But only few species have been explored for its chemical and biological studies. Plants contain chemical substances and most of which are non-nutritive dietary components that are beneficial to human health. These components are called phytochemicals. "Phyto" means that they are plant derivative.¹ Phytochemicals are chemical compound formed during plant's normal metabolic process. These chemicals are often referred to as secondary metabolite of which they are several classes including alkaloids, flavonoids, coumarins, steroids, glycosides, gum, phenol, tannins, terpenes and terpenoids.² Phytochemical analyses are of paramount importance for the identification of new sources of therapeutically and industrially valuable compounds with medicinal significance and for the best and most judicious use of naturally available materials.³

In humans, numerous phytochemicals have been found to be protective and preventive against many degenerative diseases and pathological processes such as: ageing, coronary heart disease, Alzheimer's disease, neurodegenerative disorders, atherosclerosis, cataracts, and inflammation.⁴ Epidemiological and clinical studies provided evidence that most of these phytochemicals exhibit their protective and disease-preventing functions through their antioxidant activities.⁵ In addition, vegetables possess compounds that are essential for their medicinal values, human productive well-being and healthy lifestyle. Most plant leaves are used as medicine while their nutritional potentials are yet to be discovered. The medicinal values of vegetables are due to their phytochemical and other chemical constituents.⁶

Recent study showed that jute leaves contain appreciable amount of micronutrients, economic value, proximate compositions and other health benefits while kenaf was reported to possess different medicinal values. For instance, Alexopoulou *et al.*⁷ reported that kenaf

leaf was applied to Guinea worms and the stem bark has been used for anaemia in Africa. In addition, Jaihyunk *et al.*⁸ reported the presence of phyto compounds in the hexane extracts of the different parts of the kenaf plant by GC-MS analysis. This publication suggested that phytol and linolenic acid content of kenaf leaf and stem may be responsible for its medicinal properties. Furthermore, in Ayurvedic medicine, the kenaf leaves are used for bilious, blood, diabetes, coughs and throat disorders.^{7,9,10} Jute leaves (locally called Ewedu by Yorubas) is commonly used as vegetable in the South western part of Nigeria. Kenaf (*Hibiscus cannabinus* L.) is an annual herbaceous fibre crop and belongs to the same family of *Malvaceae* with jute mallow (*Corchorus olitorius*), cotton (*Gossypium hirsutum* L), and okra (*Abelmoschus esculentus* L).

However, in Nigeria today, kenaf has been accepted as an industrial crop only but it is disheartening to note that kenaf is grossly underutilized in the pharmacological industries as well as in food and nutrition sciences. Presently, there is paucity of information on the phytochemical compositions of kenaf in relation to jute and hence their leaves are under-utilized. It is important to assess the phytochemical compositions of jute and kenaf leaves as a gateway to popularize its medicinal and nutritional uses.

Methodology

Alkaloid determination

Five grams of each sample of kenaf leaf and jute leaf were weighed into a 250ml beaker and 200ml of 10% acetic acid in ethanol were added and covered and allowed to stand for 4hrs. This was filtered and the extract was concentrated on a water bath to one-quarter of the original volume. Concentrated ammonium hydroxide was added drop wise to the extract (and ammonium hydroxide was added drop wise to the extract) until the precipitate was completed. The whole solution was allowed to settle and alkaloid precipitated. The precipitate was washed with dilute ammonium hydroxide then filtered. The residue is

the alkaloid which was dried and weighed according to the protocol described by Harborne¹¹.

Saponin determination

The kenaf and jute leaf samples were pound differently and 20g of each were put into a conical flask and 100cm³ of 20% aqueous ethanol were added. The samples were heated over a hot water bath for 4hrs with continuous stirring at about 55°C. The mixture was filtered and the residue re-extracted with another 200ml 20% ethanol. The combined extracts were reduced to 40ml over water bath at about 90°C. The concentrate was transferred into a 250ml separation funnel and 29ml of diethyl ether was added and shaken vigorously.¹²

Tannin determination

500g of each leaf sample were weighed into a 50ml plastic bottle. 50ml of distilled water were added and shaken for 1hr in a mechanical shaker. This was filtered into a 50ml volumetric flask and made up to the mark. Then 5ml of the filtrate were pipetted out into a test tube and mixed with 2ml of 0.1m ferric in 0.1N HCl solution and 0.008m potassium ferro-cyanide. The absorbance was measured at 120nm.¹³

Glycosides determination

5g of each sample was weighed into a beaker and 100ml of distilled water were added. The sample were soaked for 3hrs and filter to get filtrate. 1ml of filtrate were pipetted into a test tube ,2ml of 3,5-DNS (dinitrosalic acid) were added and boiled in a water bath for 10-15minutes, the mixture was allowed to cool in the test tube and 10ml distilled water were added and the absorbance at 540nm for glycoside were read up. Percentage glycoside will be determined using the formula:

$$\% \text{ Glycoside} = \frac{\text{Mean absorbance} \times \text{Vol of extract} \times 100}{1000 \times \text{weight of leaf sample}} \quad (1)$$

Flavonoid determination

The total flavonoid content (TFC) content of kenaf and jute leaves was determined by Zhishen *et al.*¹⁴ 10g of each sample was extracted with 100ml of 80% methanol. About 0.2ml was added to 4 ml double-distilled water and 0.3 ml of 5% NaNO₂ to the flask. The samples were maintained for 5min, and 0.3 ml of 10% AlCl₃ was added. After 6 min, 2ml NaOH and fill double-distilled water up to 10ml. The absorbance was measured at 510nm. TFC was calculated using a calibration curve of quercetin equivalents.

Qualitative analysis

Ten grams (10g) of kenaf and jute leaves were weighed into four different labelled conical flasks. 100ml of the four different solvents (distilled water, methanol, ethanol and ethyl acetate) were poured into the four different conical flasks to extract the phytochemicals. After 24hrs, the mixtures were filtered using what man filter paper (No.1) into conical flasks. The filtrates were concentrated by placing the flasks into water bath at 100°C. The resulting filtrate were cooled to room temperature, Qualitative tests were then conducted on the cool solution as prescribed by Azubuogu¹⁵.

Preparation of wagner reagent

13g of iodine crystal and 2.0g of potassium iodide were dissolved in water in a 100ml volumetric flask and the solution was made up to 100ml.

Preparation of mayer's reagent

1.3g of mercuric chloride and 5.0g of potassium iodide were dissolved in distilled water in a 100ml volumetric flask and the solution was made up to 100ml.

Test for Alkaloids: (a) 1ml of 1% HCl was added to 3ml of each of the extracts in a test tube. The mixture was heated for 20mins in a water bath. While heating, it was shaken continuously. The mixture was cooled and filtered. The procedure was repeated with each extract. (b) 1ml of each filtrate from (a) above was added to 0.5ml of Mayer's reagent.

Observation: A creamy colour change. (c) When 1ml of each filtrate from (a) above was added to 0.5ml of Wagner's reagent.

Observation: A brown colour precipitate.

Test for saponin: (a) Frothing test: 3ml of each extract and dilute with 2ml of distilled water was added in a test tube. The mixture was shaken vigorously.

Observation: A persistent frothing was observed. The frothing was persistence in the extract.

Note: forth is a mass of small bubbles especially on the surface of a liquid.

(b) Emulsion Test: 3ml of each extract was added to 5 drop of Olive oil in a test tube and the content was vigorously shaken.

Observation: Emulsification was observed which indicates the presence of saponin.

Test for flavonoids: 3ml of each extract was added to 10ml of distilled water the solution was shaken. 1ml of 10% NaOH solution was added to the mixture.

Observation: Yellow coloration was observed. Absence of yellow coloration in the mixture indicates that flavonoid was not present.

Test for Steroids: Salkowski Test: 5 drops of concentrated H₂SO₄ were added to 1ml of each extract in a separate test tube.

Observation: A red coloration was observed indicating the presence of steroid.

Test for Tannin: 2ml of each extract in a separate test tube were boiled gently for 2min and allowed to cool. 3 drop of ferric chloride solution were added to each extract

Observation: Orange coloration was observed.

Glycosides: 1ml of aqueous extract was mixed with 1ml of 20% solution of 3,5-dinitrosalic acid in methanol and 1ml of a 5% aqueous NaOH was added.

Observation: An immediate bright orange colour was observed indication of the presence of cardenolides in the extract. The colour fades gradually through reddish brown to brownish yellow. And it indicates the presence of glycosides. Heat in boiling water to get brick red coloration.

Results and discussion

The role of extraction-solvent is important in phyto-screening. Naturally, water- soluble phytochemicals like flavonoids are best extracted with water as to retain the physicochemical properties of

phytochemicals while lipophilic phytochemicals are effectively extracted with protic solvents such as ethyl acetate among others. The results in Table 1A showed that the effect of the four solvents on the extraction of phytochemical from kenaf and jute leaves. Ethanol and methanol extracted saponin, and tannin, ethyl-acetate extracted alkaloid, saponin, glycoside and tannin, and water extracted saponin, flavonoids steroids and glycosides. In all the solvents used, only water extracted most of the phytochemicals in jute leaves, the reason may be that water is an universal polar solvent. On the other hand, the results in Table 1B showed that the effect of the four solvent on the extraction of phytochemical kenaf leaf. Ethanol and methanol extracted saponin and tannin. Ethyl-acetate extracted alkaloid, glycoside, saponin and tannin and lastly water extracted flavonoids, alkaloid, saponin and glycoside. From the result table water and ethyl-acetate extracted most of the phytochemicals in the kenaf leaf sample.

Table 1A Qualitative phytochemical screening of jute leaf extract from different solvents

Parameter/ Solvent	Distilled water	Ethanol	Ethyl-acetate	Methanol
Alkaloids	—	—	+	—
Flavonoids	+++	—	—	—
Glycoside	++	—	+	—
Saponin	+	+	++	+
Steroids	+	—	—	—
Tannin	—	+	++	+

Table 1B Qualitative phytochemical screening of kenaf leaf extract from different solvents

Parameter/ Solvent	Distilled water	Ethanol	Ethyl-acetate	Methanol
Alkaloids	+	—	+	—
Flavonoids	+++	—	—	—
Glycoside	+++	—	+	—
Saponin	+	+	+++	+
Steroids	—	—	—	—
Tannin	—	+	++	+

Key + Means present, ++ mildly present, +++ densely present, — absent

The results in Table 2 showed that the alkaloid content in kenaf leaf was significantly higher than that of jute leaf. This finding supported the publication report by Okoye & Ebeledike¹⁶ that the presence of alkaloids signified the possession of medicinal properties within the leaves. However, the saponin content of jute leaf was slightly higher than that of kenaf leaf. But the two leaves under investigation appeared to possess lower level of saponin compared to *Moringa oleifera* and *Azadirachta indica* leaves.¹⁵ Similar trend of values was observed for the tannin in which jute leaf had 0.01% while kenaf leaf had 0.003%. This result was significantly lower than 0.08% tannin reported for *Moringa oleifera* and *Azadirachta indica* leaves by Azubuogu.¹⁵ In fact, these results were in strong agreement with the data reported on the phytochemical constituents of *Piper guineense* (Uziza) by Okoye and Ebeledike.¹⁶ The outcomes of this research work uphold the assertion from the study of related literature that leaves should

continue to be used as food since it contains valuable vitamins and minerals. Similarly, jute leaf contained slightly higher flavonoids and glycoside compared to kenaf leaf. Conversely, the level of glycoside obtained in kenaf and jute leaves were significantly higher than the 0.005% glycoside recorded for *Moringa oleifera* leaves by Azubuogu¹⁵. The observation of this study supported many literatures that the extracts exhibited antibacterial activity due to the presence of tannins, saponin and alkaloids.^{8,17} Jute leaf appeared to be a rich source of flavonoids which inhibit free radical chains reactions and flavonoids possess antioxidant activity and equally anti-inflammatory and antiviral have the ability to lower cholesterol level.¹⁶ On the other hand, kenaf leaf recorded higher level of carotenoid and steroid than jute leaf. However, both jute and kenaf leaves proved to be the potential sources of carotenoids which are precursor for retinol synthesis (vitamin A). Carotenoid is also antioxidant and it plays essential role in scavenging free radicals which cause degenerative diseases. The phenolic and flavonoid compounds present in the plant tissue suggest its medicinal importance.^{10,12,18} The results of this study strongly supported the publication report by Jaihyunk *et al.*⁸ that the functional groups in phenolic and flavonoids compounds present in the kenaf plant exhibit antioxidant properties and inhibit the angiotensin I-converting enzyme and lipid peroxidation. From the results of this study, it can be concluded that kenaf and jute leaves contain phytochemicals (flavonoid, alkaloids, tannin, glycosides, carotenoid, saponin, and steroids). These leaves are rich in phytochemicals and antioxidants may be explored for their nutritional and health benefits.^{19,20}

Table 2 Comparative phytochemical content of kenaf and jute leaves

Phytochemical	Kenaf leaf	Jute leaf
% Alkaloid	0.28 ^a	0.25 ^b
% Saponin	0.15 ^b	0.16 ^a
% Tannin	0.003 ^b	0.01 ^a
% Glycoside	0.10 ^b	0.11 ^a
% Steroid	0.002 ^a	0.001 ^b
Flavonoid (mg/100g)	20.00 ^b	25.00 ^a
Carotenoid (mg/100g)	569.55 ^a	546.7 ^b

Values are means \pm SD of triplicate determination. Values with the same letter in a row are not significantly different ($P < 0.05$)

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Conflict of interest

Authors declare that there is no conflict of interest.

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