Introduction

**Tel burna**

Tel Burna is located in the Shephelah of southern Israel along the northern banks of Nahal Guvrin (Figure 1). According to archaeological survey, it seems that the site was established in the Early Bronze Age II, and settled intensively in the Middle Bronze Age, Late Bronze Age and Iron Age with more scarce activity during the Persian and Byzantine periods. The survey results have been largely corroborated by the excavations which indicate that the Late Bronze Age was one of the two periods (along with the Iron II) in which the settlement at Tel Burna reached its peak.
Archaeological context

Since 2011, excavations have been carried out on a platform (Area B1–c. 40x100m; Figure 2) located below and to the west of the summit of Tel Burna. Before we began excavating this platform, intensive archaeological survey seemed to indicate that this area was settled only in the Late Bronze Age due to the high quantities of Late Bronze Age sherds and the absence of other periods.4,5 Already in the first excavation season this determination was shown to be accurate, as we uncovered an extensive Late Bronze layer that was primarily founded on the bedrock directly beneath the top soil. After seven seasons (2011-2017), the excavations have revealed the remains of a large cultic structure with a large courtyard.4,5 The exact layout of the structure remains unclear with sections of the building suffering from severe erosion. The western wall of Building 29305 is preserved to a length of 17 meters. While our interpretation remains tentative, it seems that the building is bounded on the north by a poorly preserved wall (W734007) and on the east and south by walls (W63405 and W63408) that appear to form a corner near the base of the tell’s summit. Within the building, we found evidence of two tanurs (ovens) (L53403 and L29104), which may indicative that it was a large open air courtyard.

A mittani style cylinder seal

During the 2012 season at Tel Burna, a cylinder seal was found in Area B1. The seal is made of steatite (see below for spectrographic analysis). It measures 20 mm in height by 7 mm diameter wide (Figure 3). It shows a procession of running lions moving towards the right. The procession of animals was a common motif on seals from Kamid el-Loz. Even though there are no examples of seals with such a procession of lions, the use of the steatite and the elongated figures of lions have clear parallels with some of the cylinder seals found in Kamid el-Loz. The presence of two circle border lines on the top and bottom of the seal point towards a Mitannian influence. One should also note the uncommon vertical design. The lions are also associated to the goddess Astarte or Asherah who are often depicted as a naked female standing on the backs of lions. However, the lion is also sometimes associated with Baal/Hadad, which is the Northwest Semitic name for the Sumerian storm-god Ishkur, and is usually depicted as a ‘roaring lion’. On the other hand, the lion is also an attendant of the Storm-God in general and particularly, the Akkadian Adad or Hadad is usually accompanied by a roaring lion.
Fourier transform infra-red spectroscopy analysis

The seal was first analyzed using Fourier Transform Infra-red Spectroscopy (FTIR), in order to identify its mineralogy. A hidden surface of the seal was carefully scratched to obtain ca. 0.2mg of sample. Spectra were obtained by placing the resultant powder into an agate mortar and pestle, grinding and mixing with KBr (IR-grade) and pressing into a pellet using a hand press. Spectra were collected between 4000 and 400 cm⁻¹ at 4 cm⁻¹ resolution for 32 scans using a Thermo iS5 portable spectrometer. The advantage of this method lies in its ability to provide useful information on the mineral phase present - while only using a very small sample, within a matter of minutes. The FTIR spectrum (Figure 4) obtained from the seal along with the spectrum of standard enstatite mineral (wards). It is clearly seen that the two spectra are almost identical, identifying the seal to be made of enstatite (MgSiO₃). This mineral is the magnesium end member of the pyroxene-silicate series, and is a common rock-forming mineral in igneous and metamorphic rocks. It can appear in various grades of white, green, grey yellow or brown (depending on exact composition). By mass, enstatite is approximately 60% silica (SiO₂) and 40% magnesia (MgO) and may contain minor quantities of other oxides such as CaO or Al₂O₃, and trace to minor Fe-Cr oxides.

![Figure 4](image1.png)

Figure 4 Scanning electron microscope images, showing areas of the sample at which metallic luster was observed under optical microscopy. Noted are points of EDS analysis (see table below).

Enstatite can be synthetically formed by firing of steatite-a hydrous magnesium silicate, naturally occurring rock (also known as soapstone) made up primarily of the mineral talc. Steatite is soft and therefore easily carved and shaped, and can be also made into a paste to be used in a mold (usually by mixing with other constituents). Upon heating, steatite decomposes and recrystallizes into the more hard and durable form of enstatite. Heat-treated steatite was used from at least as early as the 7th millennium BCE for the production, of small objects such as beads, scarabs and seals. The use of glazing applied onto steatite/enstatite is also known and documented. Under optical microscopy, some surface areas of the seal show metallic copper/gold like luster (which raised the assumption that it might have been covered/gilded in antiquity). The sample was further studied using an environmental SEM equipped with an Oxford EDS analyzer, without further treatment (no surface coating) and was analyzed at low vacuum mode (Figure 4). Results show that the general matrix of the sample is indeed composed of high percentage of silica (an average of ca. 53%) whereas the average magnesia content is somewhat lower than expected (see Table 1). However, other constituents were also identified, including minor concentrations of alumina (Al₂O₃), lime (CaO), potash (K₂O), soda (Na₂O) and iron, chromium and copper oxides. In some specific points of analysis (not shown in images, Table 1), high Cr-Fe oxides content was measured—which is well known for steatite/enstatite minerals.

<table>
<thead>
<tr>
<th>Site</th>
<th>Spc 1</th>
<th>Spc 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>Site 1: spc1</td>
<td>Matrix with needles</td>
<td>1.7</td>
</tr>
<tr>
<td>Site 1: spc2</td>
<td>Matrix with square crystal</td>
<td>1.5</td>
</tr>
<tr>
<td>Site 3: spc1</td>
<td>Matrix</td>
<td>3</td>
</tr>
<tr>
<td>Site 3: spc2</td>
<td>Matrix with needles</td>
<td>2.2</td>
</tr>
<tr>
<td>Site 2: spc1</td>
<td>Chromium inclusion</td>
<td>1.7</td>
</tr>
</tbody>
</table>

Table 1 Results of SEM-EDS analysis, elements detected represented as oxides and normalized.

The relative low magnesia concentration on the one hand and the high concentration of other constituents (especially lime, alumina, potash and soda) on the other may indicate that a glazing layer was applied to the seal surface. Another option is that the seal was made by mixing steatite with other constituents to make a paste and then heated. Enstatite with a small amount of iron oxides (up to 12%), like in the case of Burna seal, may exhibit a metallic sheen/luster of a bronze-like color, called bronzeite (Figure 4). This phenomenon is
known to happen upon weathering of the iron oxides which results in the separation of iron as very thin films along the cleavage/cracks of the mineral. Results show that the seal is made of enstatite, likely to be made by heating of steatite. Although copper was detected in the seal matrix, in considerable concentrations, traces of metallic copper or gold were not observed. Thus, no evidence for gilding or the application of fine metallic copper surface was found. Copper is usually a pigment used in glazes, resulting in a range of light bluish colors. No remnants of such colors were observed. Whether a glaze was applied to the surface of the seal is impossible to tell without further studying the object by creating a section. 

**Conclusion**

Under microscopic inspection, the seal showed gleaming elements that we initially thought could have been due to the presence of gold or copper, but the spectrographic analysis showed that those elements are produced by bronzite. This seal was probably included in a necklace or bracelet and that it was most likely produced outside the land of Canaan, which is not surprising due to the international commercial activity that existed in the Levant during the Late Bronze Age. It was likely used as an amulet, perhaps associated with the cult of Astarte or Asherah, consort(s) of Baal. The presence of this seal in this specific building supports the interpretation that it was a public building and in combining this find with the various imported finds that were recovered within it. Thus, it reflects the eclectic nature of the inhabitants of this building.

**Acknowledgments**

None.

**Conflicts of interest**

Authors declare that there is no conflict of interest.

**References**


