

# Development and Validation of RP-HPLC Method for the Identification of Process Related Impurities of Zolmitriptan

## Abstract

The study was focused toward synthesis, characterization and quantification of 3-Ethyl-indole impurity in Zolmitriptan formulations by Reverse Phase High Performance Liquid Chromatography method. The synthesis of a process related impurity of Zolmitriptan was successfully carried out by Fischer indole procedure. The impurity was purified by column chromatography. Characterization was done by I.R, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and GC-MS. Based on the spectral data, the structure of impurity was characterized as 3-Ethyl-indole. An efficient isocratic RP-HPLC was developed and validated according to ICH guidelines with respect to specificity, accuracy, linearity and precision. The validated HPLC method was used for detection and quantitation of 3-Ethyl-indole, a process related impurity of Zolmitriptan, from Zolmitriptan tablet formulations. The above method was found to be specific, accurate, precise, rugged and robust and can be used for routine analysis.

**Keywords:** Zolmitriptan; Fischer indole; Column chromatography; isocratic; Rugged; Robust; Specificity; Accuracy; Linearity; Precision

## Research Article

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**Abbreviations:** NMR: Nuclear Magnetic Resonance; HPLC: High Performance Liquid Chromatography; RP-HPLC: Reverse Phase High Performance Liquid Chromatography; HPTLC: High Performance Thin Liquid Chromatography; TLC: Thin Liquid Chromatography; API's: Active Pharmaceutical Ingredient's; LOD: Limit of Detection; LOQ: Limit of Quantitation; FTIR: Fourier Transform Infra Red spectroscopy

## Introduction

In Pharmaceutical World, an impurity is considered as any other organic materials, besides the drug substances, or ingredients, arises out of synthesis or unwanted chemicals that remains with Active Pharmaceutical Ingredient's (API's). The impurity may be developed either during formulation or upon aging of both API's and formulations. Presence of impurities in trace quantity in drug substance or drug product is inevitable. Therefore, their level should be controlled and monitored. They reinforce or diminish the pharmacological efficacy of the Active Pharmaceutical Ingredient's [1].

ICH defines impurities profile of a drug materials is "A description of the identified and unidentified impurities, present in a new drug substance." For Pharmaceutical products, impurities are defined as "substance in the product that are not the API itself or the excipient used to manufacture it" i.e. impurities are unwanted chemical that remains within the formulation or API in small amounts which can influence Quality, Safety and Efficacy, thereby causing serious health hazards [2].

Qualification of the impurities is the process of acquiring and evaluating data that establishes biological safety of an individual

impurity; thus, revealing the need and scope of impurity profiling of drugs in pharmaceutical research. Identification of impurities is done by a variety of Chromatographic and Spectroscopic techniques, either alone or in combination with other techniques [3-5]. There are different methods for detecting and characterizing impurities with TLC, HPTLC, and HPLC etc. Conventional Liquid Chromatography, particularly, HPLC has been exploited widely in field of impurity profiling; the wide range of detectors, and stationary phases along with its sensitivity and cost effective separation have attributed to its varied applications. Various regulatory authorities like ICH, USFDA, Canadian Drug and Health Agency are emphasizing on the purity requirements and the identification of impurities in Active Pharmaceutical Ingredient's (API's) [6-8]. According to ICH guidelines on impurities in new drug products, identification of impurities below the 0.1% level is not considered to be necessary, unless potential impurities are expected to be unusually potent or toxic. According to ICH, the maximum daily dose qualification threshold is considered as follows;  $\leq 2\text{g/day } 0.1\%$  or  $1\text{ mg per day intake (whichever is lower)} \geq 2\text{g/day } 0.05\%$  [9-12].

## Materials and Methods

### Materials reagents and chemicals

Butanaldehyde, silica gel, hydrochloric acid etc. were purchased from Merck Chemicals Pvt. Ltd. Nasik, MS, India are of AR grade. Methanol, benzene of AR grade and the acetonitrile, methanol and water of HPLC grade were purchased from Merck Chemicals Pvt. Ltd. Nasik, MS, India. The Zolmitriptan tablet formulations of different batches were purchased from local market of Kopergaon.

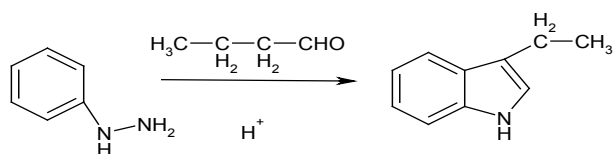
Melting points were determined by open capillary method and are uncorrected. The NMR spectra were recorded on sophisticated multinuclear FT-NMR Spectrometer model Advance-II (Bruker) using dimethylsulfoxide-*d*<sub>6</sub> as solvent and tetramethylsilane as internal standard for <sup>1</sup>H and <sup>13</sup>C NMR. IR spectra were recorded on Shimadzu FTIR-8400S spectrophotometer using KBr disc method.

### Chromatographic conditions

The quantitation of indole from formulation was carried out by HPLC method. The LC20AD Prominence Liquid Chromatography SPD20-A Shimadzu, Japan with UV-Vis detector and C18 column with dimension on 25 x 0.6cm was used for the method development with flow rate 1.0ml/min at wavelength 236nm. The methanol: acetonitrile: water in proportion of (35v:38v:27v) as a mobile phase, for development of chromatogram. The method was validation for synthesized compound and various parameters according to ICH guidelines (Q2B) were studied.

### General method for 3-ethyl-indole synthesis

Synthesis was started by Fischer indole synthesis using phenyl hydrazine with appropriate aldehyde and hydrochloric acid was refluxed in presence of methanol for 2hrs and then filtered to offer indole derivative [13].



### Preparation of standard solution

100ppm (100µg/ml) of standard solution of synthesized compound was prepared by dissolving in methanol. 10mg of standard synthesized compound was dissolved in methanol up to 100ml. From this stock, 1ml of solution was pipette out and diluted by the solvent up to 10ml to prepare 10ppm (10µg/ml) of solution

### Preparation of test solution

Test solution of formulation with different concentration from 1ppm to 10ppm was prepared using same solvent. Validation parameters were carried out for impurity by calculating range, linearity, accuracy, precision, ruggedness, robustness, LOD and LOQ as per ICH guidelines.

### Analytical method validation

A suitable analytical method was developed and validated for identification. New drug development requires meaningful and reliable analytical data to be produced at various stages of development.

### Preparation of mobile phase

The selection of mobile phase was according to polarity and non-polarity of solvents. The methanol: acetonitrile: water was selected as mobile phase in ratio of 35:38:27 and was filtered on membrane filter (0.45µ) to remove degassing and were stirred for 10-15min.

### Preparation of stock solution standard

The stock solution was prepared according to the standard procedure viz., 10mg of synthesized compound was accurately weighed on analytical balance and using mobile phase it was dissolved to make volume up to 100 ml stock solution. The sample was prepared in the ppm in the range of 1-6ppm in concentrations respectively for the method validation by HPLC.

### Preparation of sample solution (formulation)

Stock solutions of 2 different batches of Zolmitriptan marketed formulation of 100ppm in 100ml volumetric flask were prepared. For the tablet formulation 20tablets from each 2tablet batch were crushed respectively. The powder of this formulation equivalent to 10 mg of the drug was used to prepare the stock solution. Further dilute to 1ppm, 2ppm, and so on, were prepared by taking 0.1ml, 0.2ml and so on of standard test solution and diluting it to 10ml. Validation experiment was performed to demonstrate system suitability, linearity, precision, accuracy study, ruggedness and robustness as per ICH guidelines.

## Results and Discussion

### Physicochemical properties (Table 1).

Table 1: Physicochemical properties of synthesized compound.

S. No	Characteristics	Impurity	
4	Molecular formula	C <sub>10</sub> H <sub>11</sub> N	
5	Molecular weight	145	
6	Melting point	138°C	
7	Elemental Analysis	C	82.72%
		H	7.64%
		N	9.65%

### Thin layer chromatography

The sample preparation was done by dissolving solute in Methanol.

The Mobile phase Benzene: Methanol (6:1 v/v) R<sub>f</sub> value- 0.80 (Figure 1).

$$R_f = \frac{3.8}{4.7} - R_f = 0.80$$



Figure 1: TLC of Synthesized compound.

## Characterization of synthesized compound

The synthesized compound was characterized by using Advance Analytical Techniques such as UV spectra, I.R spectra, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and GC-MS. Following are the recorded spectra.

### UV- Spectra

The  $\lambda_{\max}$  of 3-ethyl-indole in methanol as solvent was found at 236nm. Since the compound substitute benzene chromophore,  $\lambda_{\max}$  confirms the structure (Figure 2).

### I.R Spectra (Figure 3).

**Spectral data:** The major functional group are primary amine, nitro, and carbonyl group obtained peak in IR spectra are as follows.

#### IR (KBr) cm<sup>-1</sup>

3341 (NH- stretching), 2850 (aliphatic -CH stretching), 2921, 2853 (aromatic C-H stretching), 1368 (C-H bending), 1297 (CH<sub>3</sub> bending), 744,750,766 (benzene ring bending). The spectral data confirm the structure of compound.

#### NMR- Spectra

**<sup>1</sup>H NMR:** For recording the <sup>1</sup>H NMR spectra of 3-ethyl-indole, the synthesized compound was dissolved in Deuteriated Dimethyl Sulphoxide and the spectrum was recorded (Table 2).

#### GC-MS

The Q-TOF Micro mass (YA-105) spectrometer capable of recording High Resolution Spectrum (HRMS) both in Atomic Pressure Chemical Ionization (APCI) and Electron Spray Ionization (ESI) were used for quantitation of synthesized impurities. The m/e ratio was calculated for synthesized compound. m/e for the synthesized compound was found at 145 confirms the mass of the synthesized impurity.

### UV method validation of impurity

UV Method was developed for synthesized compound was carried out as per ICH (Q2B) guidelines.

**Range:** The range was found to be 1-18µg/ml.

**Linearity:** Calibration curve data was constructed in the range of 1 to 18µg/ml. Beer's law was observed over this concentration range. The correlation coefficient (R<sup>2</sup>) was found to be 0.995. The regression equation Y= 0.054x+ 0.009 was found to be linear (Table 3 & Figure 4).

**Precision (Table 4).**

**Intraday precision (Table 5).**

**Interday precision: (Table 6).**

**Robustness (Table 7).**

**Ruggedness (Table 8 & 9).**

**Accuracy:** The accuracy with known concentration of synthesized compound in formulation was determined. The recovery assessment was performed by the analysis of Zolmitriptan formulation spiked with known amount of impurity at three concentration level in triplicate of 50%, 100%, 150% and was found to be accurate as shown in above Table 10 & 11.

### Limit of detection:

$$LOD = \frac{3.3 \times \text{Standard deviation}}{\text{Slope}}$$

$$LOD = \frac{3.3 \times 0.002819}{0.054}$$

$$LOD = 0.1722$$

### Limit of quantitation:

$$LOQ = \frac{10 \times \text{Standard deviation}}{\text{Slope}}$$

$$LOQ = \frac{10 \times 0.002819}{0.054}$$

$$LOQ = 0.5220$$

The LOD and LOQ were found to be 172.2ng and 522.0ng respectively, which indicate the sensitivity of method (Table 12).

### HPLC method validation

The ICH Q2B guidelines discuss the analytical method validation on HPLC. Currently the vast majority of process-related impurity determinations are performed by HPLC. It offered the desired sensitivity for trace level determinations with a high degree of automation. A wide variety of stationary phases and operation modes make HPLC applicable to all drug classes. The typical detection limits for process-related impurities by HPLC are 0.1% or lower and this can be routinely met in the majority of circumstances using conventional UV detectors. These methods involved the prediction of likely impurities within the synthetic process, their isolation and identification by suitable analytical techniques. The last step of the present study was to develop, validated HPLC method for detection and quantification of 3-ethyl-indole impurity in tablet formulations.

**HPLC chromatogram of zolmitriptan** (Figure 5 & Table 13).

**HPLC chromatogram of synthesized compound** (Figure 6 & Table 14).

**HPLC chromatogram of tablet** (Figure 7 & Table 15).

**Optimized chromatographic condition** (Table 16).

**Linearity** (Figure 8 & Table 17).

**Precision:** The precision of the intermediate was quantified for repeated concentration of 4µg/ml in range and was reliable with their area of chromatogram as shown in above table. The Standard deviation (SD) and Relative standard deviation (RSD) was found to be 1.331 and 0.317 respectively (Table 18).

**Intraday precision after 4hours** (Table 19).

**Interday precision after 24hours** (Table 20).

**Robustness** (Table 21).

**Ruggedness** (Table 22 & 23).

### Accuracy

Accuracy study was performed by the recovery method. The results demonstrate that the percentage recovery in tablet due to the presence of impurity in the tablet. Percentage recovery was found to be more at higher concentration level a compare to lower concentration level (Table 24).

### Limit of detection:

$$LOD = \frac{3.3 \times \text{Standard deviation}}{\text{Slope}}$$

$$LOD = \frac{3.3 \times 1.3313}{98.44}$$

$$LOD = 0.4462$$

### Limit of quantitation:

$$LOQ = \frac{10 \times \text{Standard deviation}}{\text{Slope}}$$

$$LOQ = \frac{10 \times 1.3313}{98.44}$$

$$LOQ = 0.1352$$

The LOD by HPLC was 446.2 ng and that of LOQ 135.2 ng the method is more sensitive and selective.

### System suitability parameters

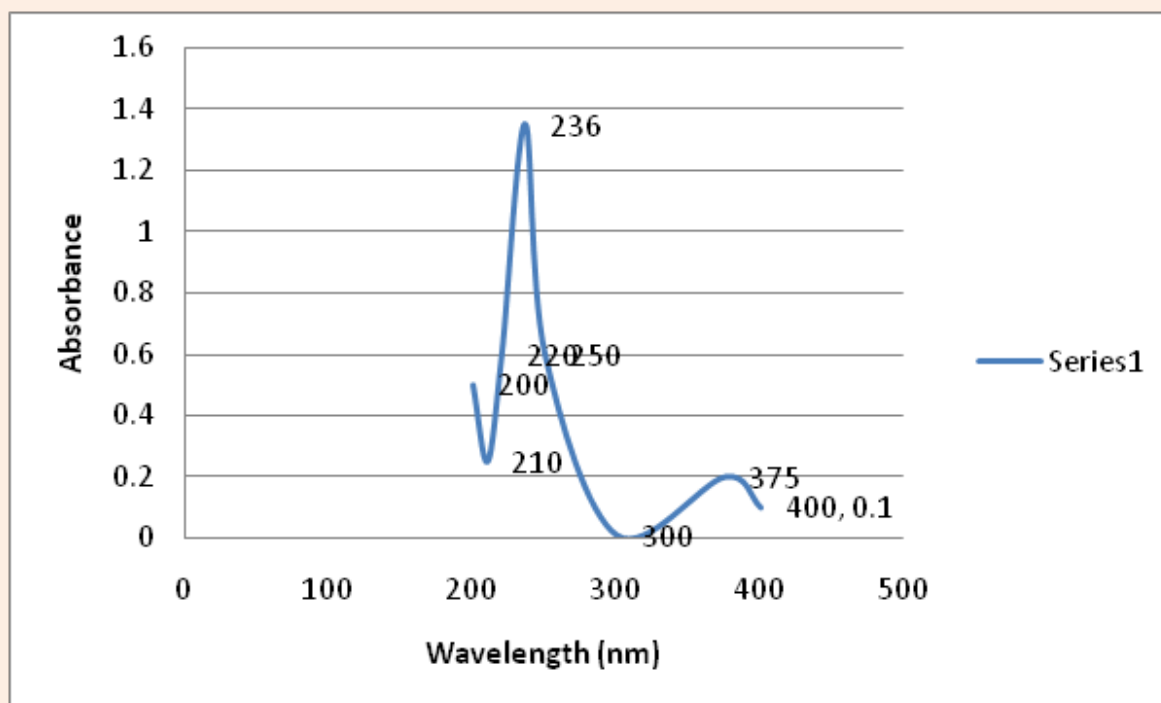
To verify that analytical system is working properly and can give accurate and precise results the system suitability parameters are to be set and it was found to be in stated range (Table 25).

### Summary of method validation parameters of HPLC (Table 26).

### Summary of retention time and asymmetry (Table 27).

### Quantitation of synthesized compound

Impurity was not found in bulk and in tablet I & II it was found to be 0.28% and 0.33% respectively. As per the ICH limit the amount of impurity is more than 0.1% indicates that the impurity found in tablet formulations is potential impurity (Table 28).



**Figure 2:** UV spectra of synthesized compound.

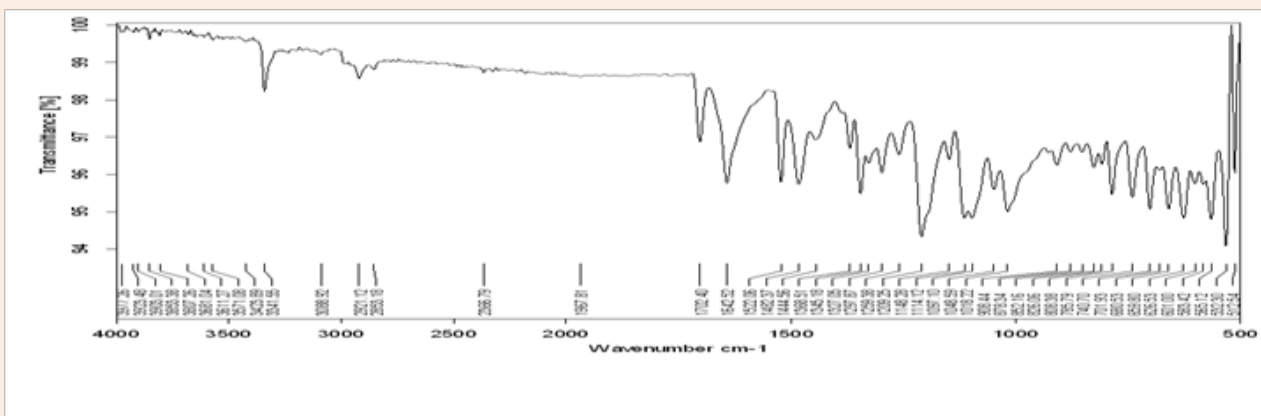


Figure 3: IR spectra of impurity 3-ethyl-indole.

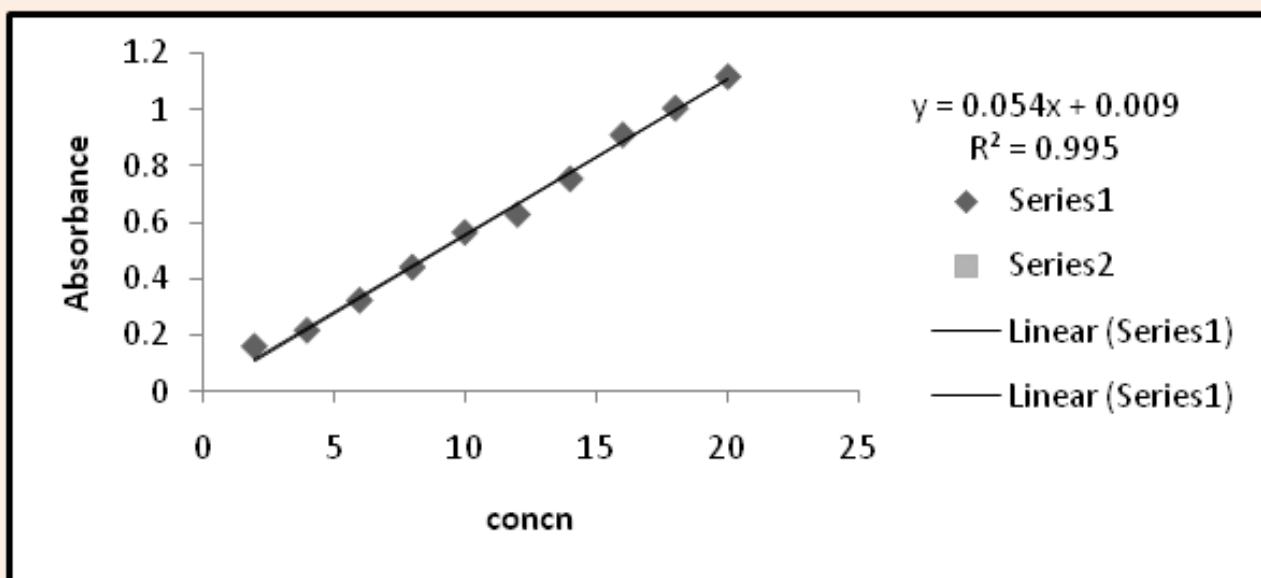


Figure 4: Graph of Linearity of synthesized compound by UV.

Table 2: <sup>1</sup>H- NMR Spectra data.

S. No	Group of Protons	Chemical shift δ (ppm)
a.	Proton attached to CH <sub>2</sub>	2.27
b.	Proton attached to CH <sub>3</sub>	3.3
c.	Alkene proton of benzene	6.4-6.8
d.	Proton of NH	5
e.	Proton of -CH of pyrrole	6.2

The PMR data supports the structural confirmation of synthesized compound 3-ethyl-indole.

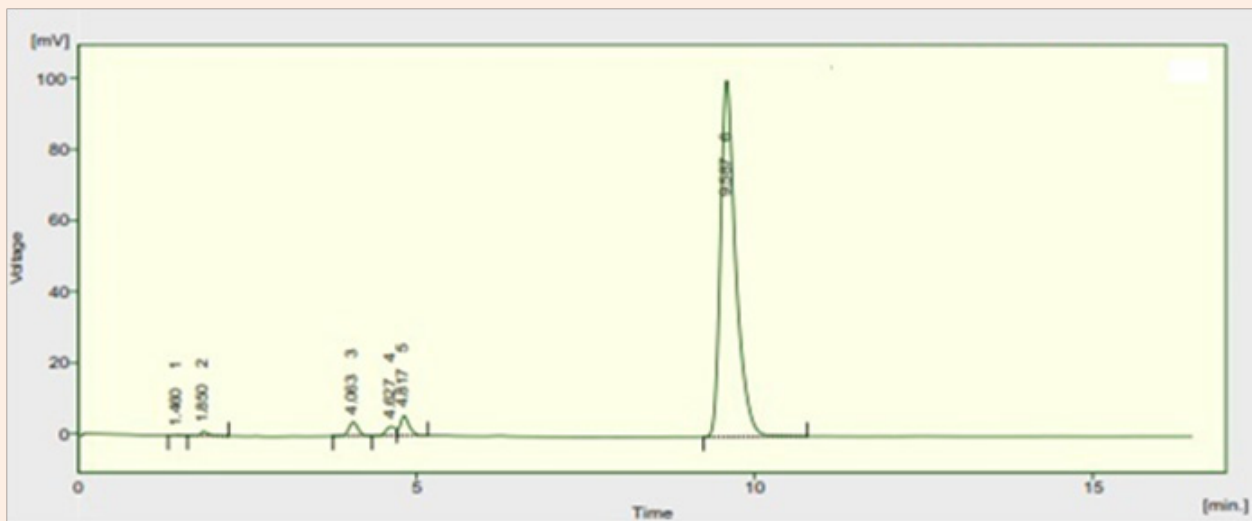


Figure 5: HPLC Chromatogram of Zolmitriptan.

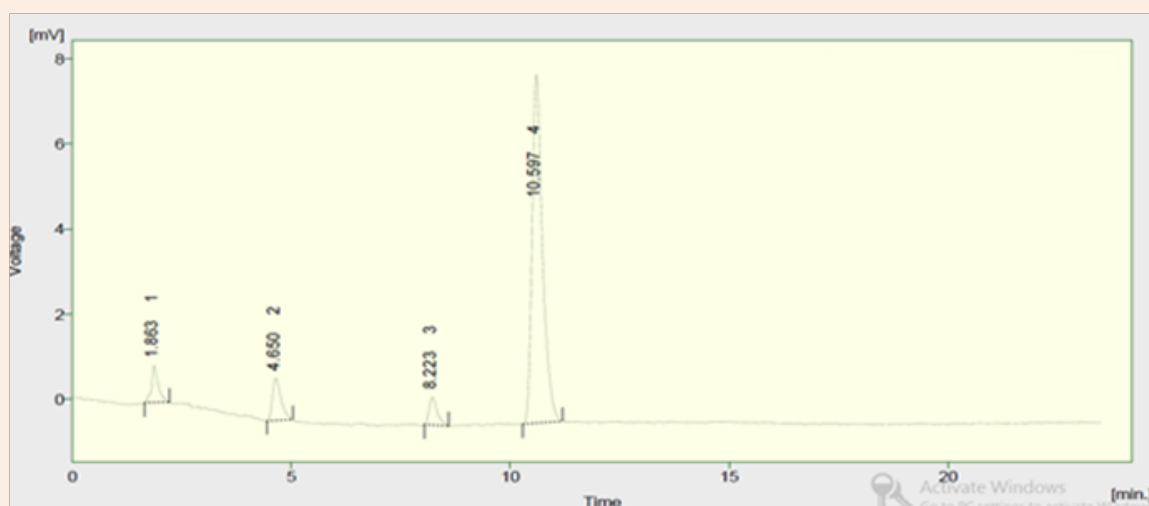
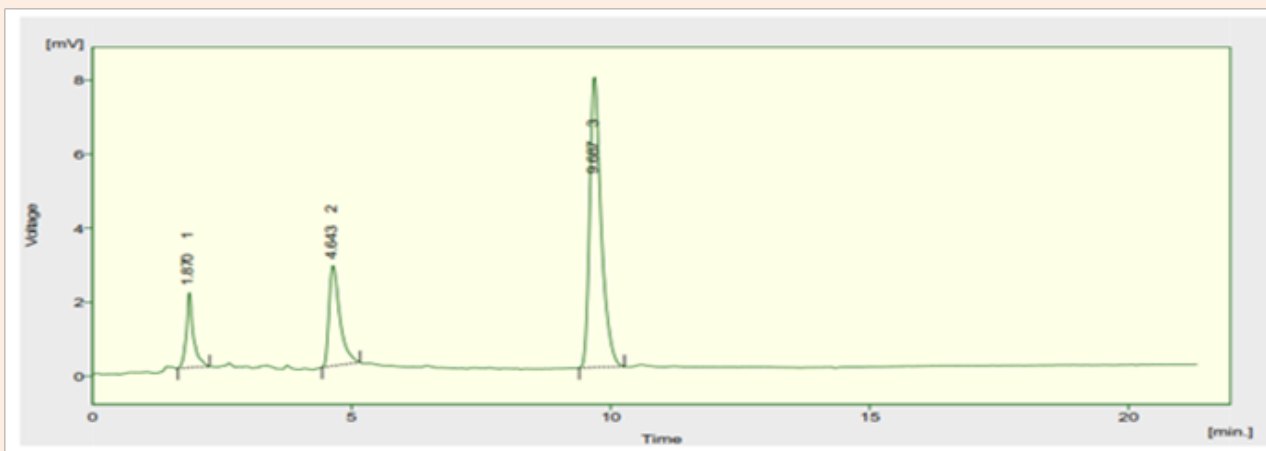


Figure 6: HPLC chromatogram of synthesized compound.

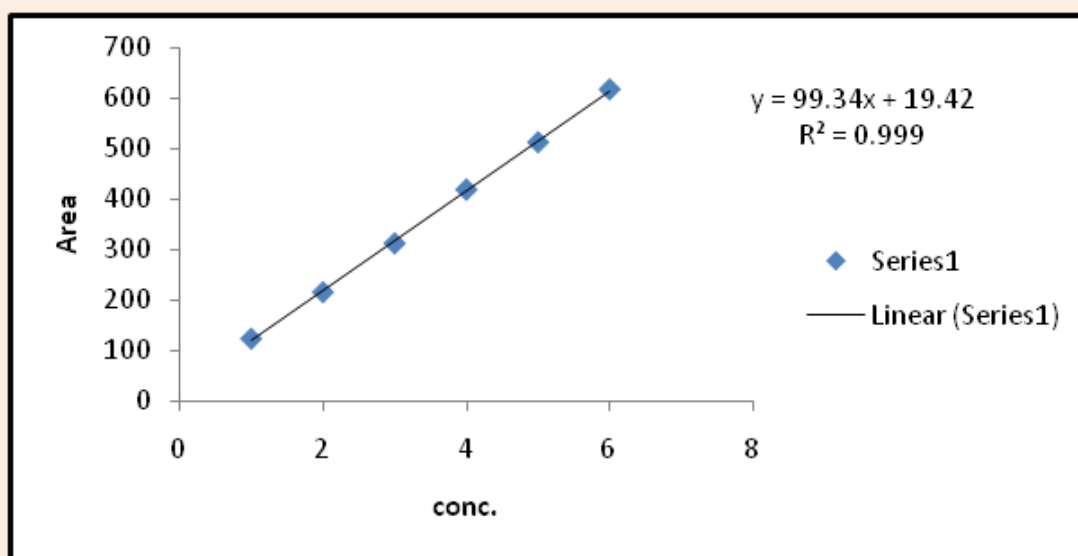
The retention time of impurity was 10.5 min and it shows a single peak which indicates purity of compound.

Table 3: Linearity of impurity by UV.

S. No	Concentration (ppm)	Absorbance at 236 nm
1	2	0.16
2	4	0.217
3	6	0.324
4	8	0.442
5	10	0.565
6	12	0.629
7	14	0.756
8	16	0.912
9	18	1.007
10	20	1.119



**Figure 7:** HPLC Chromatogram of Zolmitriptan Tablet. The retention time of Zolmitriptan tablet was found at 9.6 min.



**Figure 8:** Graph of linearity of synthesized compound by HPLC. The linearity of the proposed method was estimated by regression analysis at six concentration levels in the range of 1-6 $\mu$ g/ml for intermediate. The correlation coefficient ( $R^2$ ) was found to be 0.999 and intercept  $Y = 99.34x + 19.42$  was linear.

**Table 4:** Precision by UV.

S. No	Concentration (ppm)	Absorbance at 236nm	Mean	S.D	% RSD
1	6	0.315	0.319	0.0024	0.8777
2	6	0.321			
3	6	0.324			
4	6	0.319			
5	6	0.321			
6	6	0.318			
7	6	0.319			

**Table 5:** Intraday precision after 4 hour.

S. No	Conc. (ppm)	Abs. At 236nm after 4hr. Reading	Mean	SD	%RSD
1	6	0.309	0.316	0.0035	1.1071
2	6	0.318			
3	6	0.32			
4	6	0.318			
5	6	0.317			
6	6	0.315			
7	6	0.316			

**Table 6:** Interday precision after 24 hours.

S. No	Conc. (ppm)	Abs. At 236 nm after 24hr. Reading	Mean	S.D	%RSD
1	6	0.308	0.314	0.0032	1.1012
2	6	0.317			
3	6	0.318			
4	6	0.316			
5	6	0.315			
6	6	0.314			
7	6	0.314			

In the result of intraday and interday precision expressed by %RSD. Since variation in %RSD was not much, state the preciseness of the method.

**Table 7:** Results of robustness study by change in Instrument.

S. No	Conc. (ppm)	Abs. at 236nm (Single I)	Abs. at 236 nm (Double II)	Mean		S.D		%RSD	
				I	II	I	II	I	II
1	6	0.313	0.315	0.316	0.319	0.0031	0.0028	0.981	0.8771
2	6	0.317	0.321						
3	6	0.323	0.324						
4	6	0.316	0.319						
5	6	0.318	0.321						
6	6	0.315	0.318						
7	6	0.315	0.319						

Robustness of the method was determined by using different instrument and the respective absorbance was noted and the result was indicated by %RSD. The method was found to be robust.

**Table 8:** Result of ruggedness study by change in Analyst I and II.

S. No	Conc. (ppm)	Absorbance		Mean		S.D		%RSD	
		Analyst I	Analyst II	I	II	I	II	I	II
1	6	0.315	0.312	0.319	0.317	0.0028	0.0026	0.8777	0.8201
2	6	0.321	0.318						
3	6	0.324	0.32						
4	6	0.319	0.318						
5	6	0.321	0.317						
6	6	0.318	0.32						
7	6	0.319	0.317						

The Ruggedness was carried out by change in analyst and the difference of %RSD is negligible indicates that the method is rugged.



**Table 9:** Summary of Precision.

S. No	Parameter	SD	%RSD
1	Precision	0.0028	0.8777
2	Intraday precision	0.0035	1.1071
3	Interday precision	0.0032	1.1012
4	Robustness	0.0029	0.9293
5	Ruggedness	0.0027	0.8489

The summary of the precision is given in the above table and % RSD was found to be  $\leq 2$ .

**Table 10:** Result of recovery study by UV.

S. No	Drug/ Formulation	Percentage Recovery			Mean	S.D.	%RSD
		50%	100%	150%			
1	Tablet I	97.61	98.44	98.67	98.24	0.5575	0.5674
2	Tablet II	98.19	98.85	98.39	98.47	0.3384	0.3439

**Table 11:** Recovered amount of impurity by UV.

S. No.	Drug / Formulation	Amount of Drug ( $\mu\text{g/ml}$ )	Amount of Impurity Added ( $\mu\text{g/ml}$ )	Amount Recovered( $\mu\text{g/ml}$ )
1	Tablet I	10	5	14.63
		10	10	19.68
		10	15	24.66
2	Tablet 2	10	5	14.72
		10	10	19.77
		10	15	24.54

**Table 12:** Summary of method validation.

S. No	Parameter	Observation
1	Linearity range	1-18 $\mu\text{g/ml}$
2	Slope	0.0549
3	Intercept	0.0095
4	Correlation coefficient	0.995
5	LOD	0.17227
6	LOQ	0.52203

The developed UV method was validated as per ICH guidelines and the method is simple, cost effective to determine 3-ethyl-indole impurity in tablet formulation of Zolmitriptan for routine analysis.

**Table 13:** The Retention time of Zolmitriptan was 9.5 min.

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]	Height [%]	W05 [min]
1	1.46	2.424	0.289	0.1	0.3	0.13
2	1.85	11.705	1.278	0.7	1.1	0.11
3	4.063	37.267	3.919	2.2	3.4	0.15
4	4.627	25.732	2.608	1.5	2.3	0.17
5	4.817	52.839	5.63	3.1	4.9	0.14
6	9.587	1568.718	100.231	92.3	88	0.23
	Total	1698.684	113.955	100	100	

**Table 14:** HPLC chromatogram of synthesized compound.

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]	Height [%]	WO5 [min]
1	1.863	9.288	0.859	5.6	8	0.15
2	4.65	13.445	0.993	8.1	9.3	0.21
3	8.223	8.412	0.653	5.1	6.1	0.19
4	10.597	135.253	8.174	81.3	76.5	0.25
	Total	166.397	10.678	100	100	

The retention time of impurity was 10.5 min and it shows a single peak which indicates purity of compound.

**Table 15:** HPLC Chromatogram of Zolmitriptan Tablet.

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]	Height [%]	WO5 [min]
1	1.87	19.233	2.024	10.7	16.1	0.12
2	4.643	39.227	2.71	21.9	21.6	0.22
3	9.687	120.869	7.84	67.4	62.4	0.23
	Total	179.33	12.573	100	100	

The retention time of Zolmitriptan tablet was found at 9.6 min.

**Table 16:** Optimized chromatographic condition for RP-HPLC.

Chromatographic Conditions	Shimadzu HPLC System
Mobile phase	Methanol: Acetonitrile: Water (35:38:27)
Column	ARP-C18 (250 mm X 4.6 mm), 5 $\mu$ column
Flow rate	1ml/min
Wavelength detection	236nm
Injection volume	20 $\mu$ l
Temperature	Ambient
Retention time	10.5min
Run time	15min

**Table 17:** Result of Linearity by HPLC (Peak area vs. Conc.).

S. No	Concentration (ppm)	Area (mill volts) at 236nm
1	1	123.17
2	2	215.68
3	3	312.27
4	4	419.19
5	5	513.62
6	6	618.54

**Table 18:** Precision by HPLC.

S. No	Concentration (ppm)	Peak area (mV) at 236nm	Mean	SD	%RSD
1	4	419.19	419.89	1.331	0.317
2	4	418.98			
3	4	421.9			
4	4	419.9			
5	4	418.39			
6	4	421.01			

**Table 19:** Result of Intraday precision after 4 hours.

S. No	Conc. (ppm)	Peak Area after 4hour at 236nm	Mean	S.D	%RSD
1	4	418.83	417	2.25	0.539
2	4	413.15			
3	4	418.39			
4	4	417.12			
5	4	415.86			
6	4	418.99			

**Table 20:** Intraday precision after 24hours.

S. No	Conc. (ppm)	Peak Area after 24hour at 236nm	Mean	S.D	%RSD
1	4	423.12	424.42	1.76	0.414
2	4	422.6			
3	4	426.66			
4	4	424.24			
5	4	423.42			
6	4	426.53			

The intra and interday precision was carrying out and difference in %RSD was found not much varies and remains less than 2% indicate preciseness of method.

**Table 21:** Results of Robustness study by change in flow rate.

S. No	Conc. (ppm)	Peak Area (mV) 0.8ml/min	Mean	S.D	%RSD
1	4	787.12	789.07	3.122	0.395
2	4	784.96			
3	4	793.21			
4	4	786.68			
5	4	791.48			
6	4	789.07			

At flow rate of 0.8 ml/min

The robustness of the Intermediate was performed for change in flow rate upto 0.8 ml/min and method was robust with standard deviation 3.122 and relative standard deviation 0.395.

**Table 22:** Results of Ruggedness study by change in analyst.

S. No	Conc. (ppm)	Peak Area in mV		Mean		S.D		%RSD	
		Analyst I	Analyst II	I	II	I	II	I	II
1	4	419.19	418.8	419.94	420.66	1.392	1.481	0.3314	0.352
2	4	418.98	420.18						
3	4	421.9	420.36						
4	4	419.86	421.79						
5	4	418.39	419.93						
6	4	421.34	422.98						

The ruggedness of the Intermediate was carried out for change in analyst and method was found to be robust.

**Table 23:** Summary of Precision.

S. No	Parameter	SD	%RSD
1	Precision	1.331	0.317
2	Intraday precision	2.25	0.539
3	Interday precision	1.76	0.414
4	Robustness	3.122	0.395
5	Ruggedness	1.436	0.341

The summary of the precision is given in the above table and % RSD was found to be  $\leq 2$ .

**Table 24:** Result of recovery study by HPLC.

S. No	Drug / Formulation	Percentage Recovery			Mean	S.D.	%RSD
		50%	75%	100%			
1	Tablet I	99.22	101.3	103.79	101.43	2.288	2.255
2	Tablet II	99.25	101.68	103.13	101.3	1.969	1.934

**Table 25:** System Suitability Parameters of synthesized compound.

Property	Values	Required Limits
Retention time ( $t_R$ )	10.59	RSD $\leq 1\%$
Theoretical plates (N)	10224	$N \geq 2000$
Resolution (R)	6.36	$R \geq 2$

To verify that analytical system is working properly and can give accurate and precise results the system suitability parameters are to be set and it was found to be in stated range.

**Table 26:** Summary of Method Validation Parameters of HPLC.

S. No	Parameter	Observation
1	Linearity range	1-6 $\mu\text{g/ml}$
2	Slope	98.44
3	Intercept	21.54
4	Correlation coefficient	0.999
5	LOD	0.4462
6	LOQ	0.1352

**Table 27:** Summary of retention time and asymmetry.

S. No	Compound	Retention Time	Asymmetry
1	Zolmitriptan	9.5	1.8
2	Impurity	10.5	1.81
3	Zolmitriptan Tablet	9.6	1.78

**Table 28:** Quantisation of process related impurity of Zolmitriptan in tablets.

S. No	Formulation	Quantisation of Impurity
1	Zolmitriptan tablet I	0.28%
2	Zolmitriptan tablet II	0.33%

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